Original Research

# Estimation of Rainwater Quality Using GPS-derived Atmospheric Propagation Delay and Meteorological Data

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> Received: 29 November 2015 Accepted: 6 February 2016

# Abstract

We investigated GPS-derived atmospheric propagation delays and meteorological data for estimating rainwater quality by creating a regression model using the combinatorial group method of the data-handling algorithm (COMBI GMDH). The dependent variable was the daily wet deposition, while the independent variables were zenith hydrostatic delay, zenith wet delay, daily rainfall, and daily average wind speed. The model had a coefficient of determination ( $R^2$ ) of 0.70 and a correlation coefficient of 0.84. The mean absolute error (MAE) was 30.20, µmol/m<sup>2</sup>-day and the root mean square error (RMSE) was 40.94 µmol/m<sup>2</sup>-day. Accuracy testing to validate the model revealed an  $R^2$  of 0.95 with a correlation coefficient of 0.98. The MAE was 12.14 µmol/m<sup>2</sup>-day and the RMSE 15.35 µmol/m<sup>2</sup>-day. Rainwater quality could be estimated using GPS-derived atmospheric propagation delay (APD) and meteorological data.

Keywords: acid rain, pollution, rain water monitoring, COMBI GMDH

# Introduction

Industrial development has released pollutants into the environment everywhere on the globe that adversely affect soil, water, and air. Changes that occurred through contaminants in both the local ecosystem and the biosphere have contributed to environmental hazards such as pollution and global climate change [1]. Since the industrial revolution 200 years ago, the human collective has resulted in waste gas emissions from fossil fuel combustion (i.e., from coal, oil, and natural gas). This has released nitrogen oxides (NOx), sulfur oxides (SOx) and carbon dioxide (CO<sub>2</sub>) into the environment [2]. The emission of CO<sub>2</sub> and other gases (viz., CH<sub>4</sub>, N<sub>2</sub>O, HFCs, PFCs, SF<sub>6</sub>) have caused a greenhouse effect, and both NOx and SOx are also detrimental to health – these gases cause acid rain, which destroys the freshwater ecosystem and diminishes its recreational value [3]. Fossil fuel energy consumption is a major factor affecting our environment.

Rainwater is one of the most important natural resources [4], providing moisture to nourish humans, animals, and plants; however, with increasing air pollution from fossil fuel combustion, rainwater is contaminated and acidified by sulfuric and nitric acid [4]. The forests and watersheds are damaged [5] and nutrients in the soil eroded [6]. This has an adverse effect on livelihoods, even

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on residential buildings and historic sites [7]. As rainwater quality declines, it will no longer be fit for consumption or agriculture [8, 9].

Responsible environmental management entails the monitoring, analysis and control of acid rain, which require expertise, time, and money. There is a low pollution concentration in rain, so in order to obtain accurate and meaningful results a laboratory with advanced equipment is necessary; particularly to determine the ionic charge in the water using ion chromatography techniques. The current research proposes (a) a GPS-derived atmospheric propagation delay plus meteorological data for estimating wet deposition and (b) a combinatorial group method for data handling (COMBI GMDH) through computational modeling.

#### **Materials and Methods**

Our research comprised five steps: 1) rainwater sampling, 2) rainwater quality and wet deposition analysis, 3) GPS data recording, 4) analyzing the GPS-derived atmospheric propagation delay, and 5) regression analysis of the COMBI GMDH.

#### Rainwater Collection

Rainwater was collected (in a wet-only 200 mm diameter collector) every 24 h between 10 September 2013 and 9 September 2014 at Khon Kaen University (16°28'19.8''N, 102°49'30.7"E). Samples were stored in polyethylene bottles and stored at 4°C. To prevent contamination, the sample collector was cleaned with deionized water and new bottles were used every day.

#### Analyzing Rainwater Quality and Wet Deposition

Each rainwater sample was weighed and the quality analyzed as per standard methods [10]. pH was measured using an electrical approach (part 4500-H<sup>+</sup> B.), and the quantities of Ca<sup>2+</sup>, Mg<sup>2+</sup>, Na<sup>+</sup>, K<sup>+</sup>, and NH<sup>4+</sup> determined by ion chromatography (part 4110 C.), using a Dionex<sup>TM</sup> Ion Chromatography, Model DX-100. A Dionex<sup>TM</sup> Ion Chromatograph Model ICS-1000 analyzed the anions (viz., SO<sub>4</sub><sup>2-</sup>, NO<sup>3-</sup>, Cl<sup>+</sup>, PO<sub>4</sub><sup>3-</sup>, HCOO<sup>-</sup>, and CH<sub>3</sub>COO<sup>-</sup>).

Wet deposition analysis (acidic rain) was calculated from the wet deposition flux density; it took into account the concentration of the pollutants in the precipitation, precipitation volume, and the catchment size of the collector (equation below):

$$F = \frac{C * V}{A}$$
(1)

...where 'F' is the daily wet deposition flux ( $\mu$ mol/m<sup>2</sup>-day), 'C' the concentration of the pollutants ( $\mu$ mol/L), 'V' the daily precipitation volume (L/day), and 'A' the

catchment size of the rainwater collector (m<sup>2</sup>). The daily precipitation volume was calculated from:

$$V = \frac{m}{\rho_w}$$
(2)

...where 'm' is the precipitation mass (g) and ' $\rho_w$ ' the water density with a value of 999.97 g/L.

# Recording GPS Data

GPS data were recorded between 10 September 2013 and 9 September 2014 with a Trimble dual frequency GPS ground receiver, Model 4000SSi installed at Khon Kaen University. The geographical location was 16°28'22"N, 102°49'22"E (Fig. 1). GPS data were recorded every 30 seconds. The binary data from the GPS data were decoded and recorded in receiver independent exchange format or RINEX, using COM2RNX software, then transmitted to an IBM file-server every 24 h.

# Analyzing the GPS-Derived Atmospheric Propagation Delay

The recorded GPS data were analyzed by postprocessing to estimate the GPS-derived atmospheric propagation delay using the double difference method (DD)–GAMIT version 10.5, with software C-shell scripts developed by the Massachusetts Institute of Technology [11].

The value of the GPS-derived atmospheric propagation delay estimated from the GAMIT software was the delay in the GPS signal transmission through the atmosphere, transmitted with a dual-frequency microwave signal from the GPS satellite to the ground base. This delay is known as zenith total delay (ZTD) or zenith path delay (ZD), which can vary due to refraction of gas, water vapor, and other particles in the atmosphere [12]. The formula is:

$$ZTD = \int_{ground}^{TOA} N(z)DZ$$
(3)

...where 'N' is total refraction, 'z' the geometric height, and 'TOA' the top limit of the atmosphere. Considering the gas elements in the atmosphere, ZTD can be divided into two sub-delays: 1) the zenith hydrostatic delay (ZHD) – refraction of dry gas in the atmosphere – and 2) the zenith wet delay (ZWD) – refraction through water vapor in the atmosphere [13],

$$ZTD = ZHD + ZWD$$
 (4)

The ZHD of the GPS was the major element of ZTD with a value of 80-90%, depending on the season, region, and location of the base station. The value was estimated from the surface pressure, latitude, and height of the station, as defined by:

ZHD = (2.2779 ± 0.0024) 
$$\frac{P_s}{f(\phi, H)}$$
 (5)

...where 'P<sub>s</sub>' is the surface pressure (mbar), ' $\phi$ ' the latitude of the base station, and 'H' the ellipsoid height of the ground station (km). The function  $f(\phi,H)$  was calculated from Equation 6.

$$f(\phi, H) = f(\phi, H) = (1 - 0.00266\cos 2\phi - 0.00028H)$$
(6)

When ZHD is calculated, ZWD is found by deducting the ZHD value from ZTD, as estimated using the GAMIT software.

# Regression Analysis by Applying COMBI GMDH

We analyzed the relationship between GPS-derived atmospheric propagation delay and meteorological data estimating wet deposition, based on the mathematical relationship between the target variables, daily wet



deposition flux (Section 2.2), the independent variables ZHD and ZWD (Section 2.4), and the meteorological data measured in the study areas (including daily rainfall and daily average wind speed). The data were divided into two groups: 1) 90% of the data were used as a training data set for regression analysis using COMBI GMDH; and 2) the remaining 10% for testing the data set in the validation model – using GMDH Shell for Data Science 3.7.1.

GMDH is a data mining computational method. It can create a relatively accurate mathematical model without defining the relationship between the variables. The basic connection between inputs and an output can be expressed as:

$$y = a_0 + \sum_{i=1}^n a_i f_i \tag{7}$$

...where 'a' are coefficients, 'n' the number of base functions, and 'f' the base function. The well-known base function of GMDH is the Kolmogorov-Gabor polynomial:

$$y = a_0 + \sum_{i=1}^m a_i x_i + \sum_{i=1}^m \sum_{j=1}^m a_{ij} x_i x_j + \sum_{i=1}^m \sum_{j=1}^m \sum_{k=1}^m a_{ijk} x_i x_j x_k$$
(8)

...where 'x' are the input variables and 'm' the number of variables. GMDH hereby "selects" the various algorithms (i.e., combinatorial (COMBI), multilayer algorithm, harmonic and fuzzy). GMDH can, moreover, be used in neural networks – i.e., polynomial neural networks (PNN) or GMDH-type neural networks [14]. Despite requiring a longer calculation time and greater processing power, COMBI GMDH is a popular and comparatively accurate regression algorithm for analyzing not more than 25 variables [15]. The current study had only five variables and a low-priced high-performance computer, so we chose to use COMBI GMDH as the main algorithm.

#### **Results and Discussion**

#### Rainwater Quality

Eighty-five samples of rainwater were collected. The pH of the samples ranged from 5.02 to 6.90 (average, 5.62). The respective concentration range of the cations Ca<sup>2+</sup>, Mg<sup>2+</sup>, Na<sup>+</sup>, K<sup>+</sup>, and NH<sub>4</sub><sup>+</sup> in the rainwater was 3.70-141.68, 0.00-108.64, 0.00-210.00, 0.00-253.96, and 0.00-495.49 µmol/L. The respective concentration range of the anions SO<sub>4</sub><sup>2-</sup>, NO<sup>3-</sup>, Cl<sup>-</sup>, PO<sub>4</sub><sup>3-</sup>, HCOO<sup>-</sup>, and CH<sub>3</sub>COO<sup>-</sup> was 1.44-84.90, 1.95-252.34, 1.23-186.88, 0.00-14.03, 0.00-3.88, and 0.00-7.39 µmol/L. The rank of the average charge most frequently found was NH<sub>4</sub><sup>+</sup> > Ca<sup>2+</sup> > NO<sub>3</sub><sup>-</sup> > Cl<sup>-</sup> > Na<sup>+</sup> > SO<sub>4</sub><sup>2-</sup> > K<sup>+</sup> > Mg<sup>2+</sup> > HCO<sub>2</sub><sup>-</sup> > PO<sub>4</sub><sup>3-</sup> > CH, COO<sup>-</sup> (Table 1).

The rainwater in the study area showed acidity, suggesting a source of pollutants, possibly from: a)

y.

Parameter	Unit	Range	Average
pН	-	5.02-6.90	5.62
Ca <sup>2+</sup>	µmol/L	3.70-141.68	30.06
$Mg^{2+}$	µmol/L	0.00-108.64	6.31
Na <sup>+</sup>	µmol/L	0.00-210.00	14.17
$K^+$	µmol/L	0.00-253.96	8.24
$\mathrm{NH_4^+}$	µmol/L	0.00-495.49	46.14
SO4 <sup>2-</sup>	µmol/L	1.44-84.90	9.96
NO <sub>3</sub> -	µmol/L	1.95-252.34	22.73
Cl	µmol/L	1.23-186.88	17.61
PO <sub>4</sub> <sup>3-</sup>	µmol/L	0.00-14.03	0.62
HCOO-	µmol/L	0.00-3.88	0.68
CH <sub>3</sub> COO-	µmol/L	0.00-7.39	0.21

natural sources such as  $CO_2$  from respiration or NO from lightning strikes, which were dissolved in the rain, creating acid precipitation; and b) man-made sources from vehicle exhaust or waste incineration releasing NO, NO<sub>3</sub>, and/or SO<sub>2</sub>, corresponding to the number of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> charges found in the samples [5]. This human activity is primarily responsible for increasing acidifying atmospheric pollutants, which is causing acid rain. Although rainwater in the study area was acidic, no serious acid rain (i.e., a pH < 5) was observed during the research (Fig. 2).

# Wet Deposition

The respective range of wet deposition of Ca<sup>2+</sup>, Mg<sup>2+</sup>, K<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, and Na<sup>+</sup> was 0.63-173.52, 0.00-69.98, 0.00-14.07, 0.00-123.50, and 0.00-67.28  $\mu$ mol/m<sup>2</sup>-day, and of CH<sub>3</sub>COO<sup>-</sup>, HCOO<sup>-</sup>, Cl<sup>-</sup>, SO<sub>4</sub><sup>-2-</sup>, NO<sub>3</sub><sup>-</sup>, and PO<sub>4</sub><sup>-3-</sup> was 0.00-0.39, 0.00-2.25, 0.19-31.72, 0.34-93.73, 0.14-62.71, and 0.00-17.16  $\mu$ mol/m<sup>2</sup>-day (Table 2). The respective daily maximum, minimum, and average wet depositions were 431.41, 3.10, and 68.60  $\mu$ mol/m<sup>2</sup>-day (Fig. 3). Ca<sup>2+</sup> had the highest annual wet deposition by charge at 1,933.59  $\mu$ mol/m<sup>2</sup>-year, followed by NH<sub>4</sub><sup>+</sup>, SO<sub>4</sub><sup>-2-</sup>, NO<sub>3</sub><sup>-</sup>, Mg<sup>2+</sup>, Cl<sup>-</sup>, Na<sup>+</sup>, K<sup>+</sup>, PO<sub>4</sub><sup>-3-</sup>, HCOO<sup>-</sup>, and CH<sub>3</sub>COO<sup>-</sup> at 908.93, 731.92, 416.04, 284.25, 265.14, 221.05, 88.24, 73.42, 15.81, and 1.16  $\mu$ mol/m<sup>2</sup>-year, respectively.

The analysis of annual wet deposition showed that the area had particles of limestone (i.e., CaCO<sub>3</sub> and CaO) in the atmosphere: these particles generated Ca<sup>2+</sup> with the highest annual wet deposition. Alkaline dust in the atmosphere was, however, beneficial as it helped to reduce  $H_2SO_4$  (the reaction of  $H_2O$  with  $SO_2$ )[16]. The most important discovery was finding high  $NH_4^+$  in the wet deposition. If  $NH_4^+$  falls to the ground, microbes convert it to  $NO_3^-$  with the nitrification process, resulting in a higher deposition of  $NO_3^-$  and a greater possibility of  $HNO_3$ .



Date

12/6/2014

1/5/2014

Fig. 2. pH of samples.

10/9/2013

Since  $NO_3^-$  and  $NH_4^+$  are key factors in eutrophication, environmental monitoring is necessary to balance  $NH_4^+$ from fertilizer with residual  $NH_4^+$  in the soil, water, and atmosphere [16]. Importantly, the detection of high  $Ca^{2+}$ and  $NH_4^+$  was found in the agricultural zones of Thailand (i.e., Nakorn Pathom) [17]. Annual wet deposition of  $SO_4^{-2-}$  and  $NO_3^-$  were high compared to the other charges; perhaps because of combustion, motor vehicle exhaust, and incineration at nearby factories.

2/10/2013

#### **GPS-Derived** Atmospheric Propagation Delay

Based on the estimated value of GPS-derived atmospheric propagation delay of both parts, ZHD and

Ion	Range	Average
Ca <sup>2+</sup>	0.63-173.52	26.86
Mg <sup>2+</sup>	0.00-69.98	3.95
Na <sup>+</sup>	0.00-67.28	3.07
$K^+$	0.00-14.07	1.23
$\mathrm{NH_4^+}$	0.00-123.50	12.62
SO <sub>4</sub> <sup>2-</sup>	0.34-93.73	10.17
NO <sub>3</sub> -	0.14-62.71	5.78
Cl	0.19-31.72	3.68
PO <sub>4</sub> <sup>3-</sup>	0.00-17.16	1.02
HCOO-	0.00-2.25	0.22
CH <sub>3</sub> COO <sup>-</sup>	0.00-0.39	0.02

Table 2. Wet deposition (µmol/m<sup>2</sup>-day).

ZWD, the average ZHD was 2,247.29 mm. The highest value was 2,274.01 mm on 18 January 2014, and the lowest 2,227.54 mm on 12 June 2014. The average ZWD was 294.48 mm. The highest value was 417.16 mm on 19 September 2013 and the lowest was 71.00 mm on 23 January 2014 (Fig. 4). In December 2013 and May 2014 ZHD and ZWD values could not be estimated because of recorder malfunction.

24/7/2014

Comparing ZHD to ZWD over the same period, ZHD showed an inverse relationship with ZWD; ZHD increased in the dry season (November-April) and decreased during the rainy season (May-October). Conversely, ZWD decreased in the dry season and increased during the rainy season. This result agrees with that of Bevis et al., who demonstrated that ZWD was directly related to the total column water vapor, known as precipitable water vapor (PWV). When the atmosphere contained large quantities of water vapor during the rainy season, ZWD was high. Satomura et al. found that PWV in Thailand always increases during the rainy season [18]. Thus, when PWV was high in the rainy season, ZWD was also high due to the change of PWV.

#### Wet Deposition Model

Our current research covers regression by applying COMBI GMDH to analyze the relationship between rainwater quality, GPS-derived atmospheric propagation delay, and meteorological data. The target variable was daily wet deposition and the independent variables were ZHD, ZWD, daily rainfall, and daily average wind speed. Eighty-five samples were collected in order to analyze wet deposition as input data. GPS data to estimate ZHD and ZWD was only recorded on 72 rainy days, so analysis of the regression used only 72 sets of data. The data were

26/8/2014



Fig. 3. Daily wet deposition.

divided into two groups: 65 training data and seven testing data (as described in section 2.5). The regression analysis was performed by the mathematical relation in the polynomial:

$$F = 42.442 - 75.653 * ZWD * ZHD^{-1} + 0.133 * * ZWD * R + 3.384e^{6} * R * ZWD^{-2} -12.024 * R^{-1}W^{-2} - 72.886R - 1.0976R^{-1} (9)$$



Fig. 4. (a) Zenith hydrostatic delay and (b) zenith wet delay.

...where 'F' is daily wet deposition ( $\mu$ mol/m<sup>2</sup>-day), ZHD the zenith hydrostatic delay (mm), ZWD the zenith wet delay (mm), 'R' the daily rainfall (mm), and 'W' daily wind speed (m/s). This model had a coefficient of determination ( $R^2$ ) at 0.70, with the correlation coefficient at 0.84. MAE was 30.20  $\mu$ mol/m<sup>2</sup>-day and RMSE was 40.94  $\mu$ mol/m<sup>2</sup>-day. Testing the model determined  $R^2$  at 0.95, with the correlation coefficient at 0.98. MAE was 12.14  $\mu$ mol/m<sup>2</sup>-day and RMSE was 15.35  $\mu$ mol/m<sup>2</sup>-day (Fig. 5).

The  $R^2$  of the mathematical model – applied ZHD, ZWD, daily rainfall, and daily average wind speed for estimating daily wet deposition from the regression analysis was done using the COMBI GMDH. The process of creating and testing the model was successful; there was a high correlation coefficient, suggesting a close relationship between the four independent variables and daily wet deposition. This might be because the pollutants that cause contamination in rainwater (i.e., gases, sand, dust, aerosol, and ash) were the reasons for the GPSderived delay. The microwave signal - broadcast from GPS satellites - might be absorbed by atmospheric gases (e.g., water vapor, ozone, carbon monoxide, nitrous oxide, chlorine monoxide, oxygen, and ammonia). As a consequence, the ZHD value was related to the refraction index (N) and changed due to the quantities of gas in the atmosphere that caused the acid charges (i.e., N<sub>2</sub>,  $NO_2$ -caused  $NO_3^-$ , and  $H_2S$ - and  $SO_2$ -caused  $SO_4^{2-}$ ). Daily wet deposition was then calculated from the wet deposition flux (F) as directly related to rainfall (R) (Equation 1). ZWD was also related to wet deposition, as ZWD is widely accepted for estimating PWV - the parameter of water vapor in the atmosphere associated with rainfall (R) [19]. Wind speed data (W) is possibly related to rain, since wind is a significant variable for moistness [20], and a factor in moving the dust from the ground to the atmosphere [21].



Fig. 5. Plot of daily wet deposition model.

Results of regression model testing indicated the possibility of implementing the GPS-derived atmospheric propagation delay and meteorology data to estimate daily wet deposition as the  $R^2$  was acceptable and the model had a high-value correlation coefficient. The model also had a low MAE value in the creating and validation process; thus, it could be used to estimate rainwater quality in the study area quickly and efficiently. In future, if the input data (i.e., rainwater quality, meteorological data, and GPS data) were collected in high quantity and continuously, then the model might be even more accurate as the COMBI GMDH will learn and create the best model fit. The model will then be able to adjust to suit the situation and provide results closer to the actual recorded measurements.

#### Conclusions

Based on collection and analysis of rainwater at Khon Kaen University over one year, the pH of 85 samples varied between 5.02 and 6.90 (with none < 5.0). There was, therefore, no risk of hazardous acid rain found.  $NH_4^+$  was the ion with the highest concentration (46.14 µmol/L), followed in lesser amounts by Ca<sup>2+</sup>, NO<sub>3</sub><sup>-</sup>, Cl<sup>-</sup>, Na<sup>+</sup>, SO<sub>4</sub><sup>2-</sup>, K<sup>+</sup>, Mg<sup>2+</sup>, HCOO<sup>-</sup>, PO<sub>4</sub><sup>3-</sup>, and CH<sub>3</sub>COO<sup>-</sup>.

Daily wet deposition varied between 3.10 and 431.41  $\mu$ mol/m<sup>2</sup>-day (average 68.60  $\mu$ mol/m<sup>2</sup>-day). The charge with the highest yearly deposition was Ca<sup>2+</sup>, followed by NH<sub>4</sub><sup>+</sup>, SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>, Mg<sup>2+</sup>, Cl<sup>-</sup>, Na<sup>+</sup>, K<sup>+</sup>, PO<sub>4</sub><sup>3-</sup>, HCOO<sup>-</sup>, and CH<sub>3</sub>COO<sup>-</sup>, in descending order.

Estimating the GPS-derived atmospheric propagation delay in the Khon Kaen University area with GAMIT yielded a ZHD value between 2,274.01 and 2,227.54 mm (average, 2,247.29 mm), and a ZWD between 71.00 and 417.16 mm (average, 294.48 mm). ZHD was inversely

related to ZWD: ZHD increased in the dry season and decreased in the rainy season.

The COMBI GMDH regression analysis between daily wet deposition and the independent variables (viz., ZHD, ZWD, daily rainfall, and daily average wind speed) had an  $R^2$  of 0.70, while the correlation coefficient was 0.84. The respective MAE and RMSE were 30.20 and 40.94 µmol/m<sup>2</sup>-day. The  $R^2$  of the test data set was 0.95, with a correlation coefficient of 0.98, with a MAE of 12.14 µmol/m<sup>2</sup>-day, and a RMSE of 15.35 µmol/m<sup>2</sup>-day. Using meteorological conditions, atmospheric gases, and other pollutants causing contamination in the rainwater (sand, dust, and aerosols) that cause GPS delay, the COMBI GMDH model can estimate preliminary rainwater quality as quickly and as efficiently as the laboratory. This would be beneficial for rainwater quality observers in areas conducting acid rain hazard prevention measures.

#### Acknowledgements

We thank the Research Fund of Khon Kaen University, Thailand, and the Higher Education Research Promotion, National Research University Project of Thailand, Office of the Higher Education Commission for supporting this research, plus Mr. Saman Prakarnrat for Meteorological data and Dr. Akraphon Janon and Mr. Bryan Roderick Hamman for assistance with the English-language presentation of the manuscript.

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